New Triterpenoid Saponins and Neolignans from Morina kokonorica

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Two new taraxastane-type triterpenoid saponins and two new neolignans, $(6\alpha,11\alpha)$ -6-[(2-*O*-acetyl- α -L-arabinopyranosyl)oxy]-3-oxotaraxast-20-ene-11,28-diyl diacetate (**1**), $(6\alpha,11\alpha)$ -6-[(2-*O*-acetyl- β -D-xy-lopyranosyl)oxy]-3-oxotaraxast-20-ene-11,28-diyl diacetate (**2**), (1R,2R,4E)-1,5-bis(3,4-dimethoxyphen-yl)-2-(methoxymethyl)pent-4-en-1-ol (**3**), and 1,1'-[(1E,4R,5R)-5-methoxy-4-(methoxymethyl)pent-1-ene-1,5-diyl]bis(3,4-dimethoxybenzene) (**4**), together with six known compounds were isolated from the Chinese medicinal plant *Morina kokonorica* HAO. Their structures were determined on the basis of spectral data, especially 2D-NMR and HR-ESI-MS.

Introduction. - The genus Morina (Dipsacaceae) consists of ca. 15 species growing in plateau regions of temperate zones in the world. Six species, M. chinensis, M. chlorantha, M. kokonorica, M. nepalensis var. alba, M. nepalensis var. delavayi, and M. nepalensis var. nepalensis, mainly occur in mountainous regions at the altitude between 2800 and 4900 m in the northwest and southwest of China [1]. Some species of this genus have been used as Chinese traditional medicine for the treatments of cerebral apoplexy, arthralgia, lumbago, megrim, and tumors, such as M. chinensis, M. kokanica, M. nepalensis var. alba, and M. nepalensis var. delavayi [2]. Up to now, the chemical constituents of *M. chinensis* [3], *M. nepalensis* var. alba [4], *M. kokanica* [5], *M.* longifolia [6], M. officinalis [7], and M. persica [8] were reported as triterpenoids, triterpenoid saponins, lignanoids, phenylpropanoids, caffeoylquinic acids, flavonoids, and essential oils. In this article, we present the results of our systematic investigation of M. kokonorica. Two new taraxastane-type triterpene saponins, 1 and 2, two new neolignans, **3** and **4**, together with six known compounds, 5-10, were isolated from the MeOH extract of the whole plant. Their structures were determined on the basis of spectral data, including 1D- and 2D-NMR, HR-ESI-MS, as well as IR, UV, and CD. The known compounds are elucidated as 3β -hydroxyurs-12-en-28-al (= ursolaldehyde) (5) [9][10], 3β -hydroxyurs-11-en-28,13 β -olide (6) [11], morinol G (7) [12], (+)pinoresinol (8) [13], (-)-lariciresinol (9) [14], and balanophonin (10) [15] by comparison of their physical and spectral data with those reported in the literature.

Results and Discussion. – Compound **1** was obtained as a crystalline powder. The HR-ESI-MS showed a *pseudo*-molecular ion peak at m/z 748.4633 ($[M + NH_4]^+$, $C_{41}H_{66}NO_{11}^+$; calc. 748.4630), corresponding to the molecular formula $C_{41}H_{62}O_{11}$. The IR spectrum of **1** showed absorption bands at 3435 (OH), 1735 (C=O), and 1639

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(C=C) cm⁻¹. From ¹H- and ¹³C-NMR (*Table 1*), ¹H,¹H-COSY, HSQC, and HMBC, 1D-NOE, and NOESY data, the structure of **1** was determined as $(6\alpha,11\alpha)$ -6-[(2-*O*-acetyl- α -L-arabinopyranosyl)oxy]-3-oxotaraxast-20-ene-11,28-diyl diacetate.

The ¹³C-NMR spectrum (*Table 1*) of **1** showed 41 C-atoms, including seven characteristic Me groups of a triterpenoid moiety, a CO group (δ (C) 218.5), a CH₂O group (δ (C) 61.4), two O–CH groups (δ (C) 79.8 and 72.7), two olefinic C-atoms (δ (C) 117.9 (C(21)), 139.8 (C(20))), three AcO groups, and a sugar residue. The ¹H-NMR spectrum (*Table 1*) showed the presence of six tertiary Me groups at δ (H) 0.79, 1.11, 1.22, 1.24, 1.03, and 1.60, one secondary Me group at δ (H) 0.94 (d, J = 6.6, Me(29)), an olefinic H-atom at δ (H) 5.23 (br. d, J = 6.6, H–C(21)), H-atoms of two O–CH groups (δ (H) 3.72 (br. dd, J = 10.5, 3.0), 4.92–4.94 (m)), H-atoms of a CH₂O group (δ (H) 3.77 (d, J = 10.5), 4.27 (d, J = 10.5)) and an anomeric H-atom (δ (H) 4.60 (d, J = 6.0)).

	$\delta(\mathrm{H})$		$\delta(C)$	
	1	2	1	2
$CH_{2}(1)$	$1.93 - 1.95 (m, H_a),$	$1.95 - 1.97 (m, H_a),$	42.1 (<i>t</i>)	42.2 (t)
	$2.22 - 2.24 (m, H_{\beta})$	$2.19 - 2.21 (m, H_{\beta})$		
$CH_{2}(2)$	2.76 (br. dd , $J = 11.5$, 12.6, H_a),	$2.76 - 2.78 (m, H_a),$	33.1 (t)	33.2(t)
	$2.17 - 2.19 (m, H_{\beta})$	$2.16 - 2.18 (m, H_{\beta})$		
C(3)	_	_	218.5(s)	218.6(s)
C(4)	_	-	47.5 (s)	47.6 (s)
H-C(5)	1.94 - 1.96 (m)	1.95 - 1.97 (m)	56.8(d)	56.9 (d)
H-C(6)	3.72 (br. dd , $J = 10.5$, 3.0)	3.69 - 3.71 (m)	79.8(d)	79.5 (d)
$CH_{2}(7)$	$1.51 - 1.53 (m, H_a),$	$1.51 - 1.53 (m, H_{\alpha}),$	34.4 (t)	35.1 (t)
	$1.99 - 2.01 \ (m, H_{\beta})$	$1.98 - 2.00 (m, H_{\beta})$		
C(8)	_	-	41.9 (s)	41.9 (s)
H-C(9)	1.83 - 1.85 (m)	1.84 - 1.86 (m)	51.3 (d)	51.4 (d)
C(10)	-	-	39.6 (s)	39.6 (s)
H - C(11)	4.92 - 4.94 (m)	4.97 - 4.99(m)	72.7(d)	72.8(d)
$CH_2(12)$	$1.26 - 1.35 (m, H_a),$	$1.28 - 1.32 (m, H_{a}),$	35.0 (t)	35.1 (t)
	$1.74 - 1.82 (m, H_{\beta})$	$1.76 - 1.84 \ (m, H_{\beta})$		
H - C(13)	1.80 - 1.82 (m)	1.79 - 1.81 (m)	36.6(d)	36.7 (d)
C(14)	-	_	43.0(s)	43.1 (s)
$CH_2(15)$	0.99 (br. $d, J = 8.8, H_{\alpha}$),	1.00 (br. $d, J = 8.8, H_{\alpha}$),	26.5(t)	26.5(t)
	$2.14-2.15 (m, H_{\beta})$	$2.12 - 2.14 (m, H_{\beta})$		
$CH_2(16)$	$1.55 - 1.57 (m, H_a),$	$1.56 - 1.58 (m, H_{a}),$	30.0(t)	30.1(t)
	$1.13 - 1.15 (m, H_{\beta})$	$1.14 - 1.16 (m, H_{\beta})$		
C(17)	_	-	37.1 (s)	37.2 (s)
H - C(18)	1.33 - 1.35 (m)	1.33 - 1.35 (m)	48.0(d)	48.0(d)
H - C(19)	1.66 - 1.68 (m)	1.67 - 1.69 (m)	35.4(d)	35.5 (d)
C(20)	-	-	139.8 (s)	139.7 (s)
H - C(21)	5.23 (br. $d, J = 6.6$)	5.24 (br. $d, J = 6.6$)	117.9 (d)	118.0(d)
$CH_{2}(22)$	$1.98 - 2.00 (m, H_a),$	$1.99-2.01 (m, H_a),$	41.6 (t)	41.7 (<i>t</i>)
	$2.05 - 2.07 (m, H_{\beta})$	$2.04 - 2.06 (m, H_{\beta})$		
Me(23)	1.24 (s)	1.24 (s)	31.3(q)	31.3 (q)
Me(24)	1.22 (s)	1.23 (s)	18.7(q)	18.8(q)
Me(25)	0.79(s)	0.81(s)	19.1(q)	19.1(q)
Me(26)	1.11 (s)	1.13 (s)	17.4(q)	17.4(q)
Me(27)	1.03 (s)	1.03 (s)	14.4(q)	14.4(q)
CH ₂ (28)	$3.77 (d, J = 10.5, H_a),$	$3.78 (d, J = 10.5, H_a),$	61.4 (t)	61.5 (<i>t</i>)
	$4.27 (d, J = 10.5, H_b)$	$4.28 (d, J = 10.5, H_b)$		
Me(29)	0.94 (d, J = 6.6)	0.95 (d, J = 6.6)	22.6(q)	22.7(q)
Me(30)	1.60(s)	1.62 (s)	21.4(q)	21.4(q)
sugar	Ara	Xyl	Ara	Xyl
H - C(1')	4.60 (d, J = 6.0)	4.61 (d, J = 6.3)	101.0(d)	101.7(d)
H-C(2')	4.96(t, J = 7.0)	4.78(t, J = 7.2)	72.6(d)	73.7 (<i>d</i>)
H-C(3')	3.69–3.71 (<i>m</i>)	3.54(t, J = 8.2)	71.7(d)	75.5 (d)
H-C(4')	3.89 - 3.91 (m)	3.81 - 3.83 (m)	67.0(d)	70.3 (d)
CH ₂ (5')	$3.55 (dd, J = 13.5, 4.2, H_a),$	$3.31 (dd, J = 11.7, 2.1, H_a),$	63.5 (<i>t</i>)	64.3 (<i>t</i>)
	$3.91 - 3.93 (m, H_b)$	$4.03 (dd, J = 11.4, 4.5, H_b)$		
C(1")	-	-	171.4 (s)	171.4 (s)
Me(2")	2.10 (s)	2.12 (s)	21.0(q)	21.1(q)
C(1*)	-	-	171.2 (s)	171.2 (s)
Me(2*)	2.02 (s)	2.04 (s)	21.2(q)	21.3 (q)
C(1 [#])	-	_	171.0(s)	170.0(s)
Me(2 [#])	2.00 (s)	2.00(s)	21.9(q)	21.9 (q)

Table 1. ¹*H*- and ¹³*C*-*NMR* (DEPT) *Data*^a) of **1** and **2** (CDCl₃, δ in ppm, *J* in Hz)

^a) The assignments of the signals of **1** and **2** were performed by ¹H,¹H-COSY, HMBC, and NOESY experiments.

The ¹H- and ¹³C-NMR data of **1** were similar to those of taraxast-20-en-3-one [16-18], indicating that **1** is a taraxastane-type triterpene saponin.

The ¹H- and ¹³C-NMR spectra showed signals of the sugar residue at $\delta(H)$ 4.60 (*d*, J = 6.0), 4.96 (t, J = 7.0), 3.69–3.71 (m), 3.89–3.91 (m), 3.55 (*dd*, J = 13.5, 4.2), and 3.91–3.93 (m); and at $\delta(C)$ 101.0, 72.6, 71.7, 67.0, and 63.5, which closely resembled those of arabinopyranose (Ara) [19][20]. The L-configuration for Ara was assumed from biogenetic considerations. Furthermore, the characteristic anomeric H-atom signal ($\delta(H)$ 4.60 and J(1',2') = 6.0) indicated an α -configuration in comparison with the values reported for α -L-Ara ($\delta(H)$ 4.45, J = 7) and β -L-Ara ($\delta(H)$ 5.23, J = 3) in the literature [20]. The signals for H–C(2') and C(2') were shifted downfield because of the effect of an AcO group [20]. This was supported by HMBCs from H–C(2') at $\delta(H)$ 4.96 to C(1'), C(3'), and an AcO group (C(1'') at $\delta(C)$ 171.4). ¹H,¹H-COSY correlations, between H–C(1') at $\delta(H)$ 4.60 and H–C(2') at $\delta(H)$ 4.96, and between H–C(2') and H–C(3') at $\delta(H)$ 3.69–3.71, further confirmed the position of the AcO group in the sugar residue.

The structure of **1** and its ¹H- and ¹³C-NMR assignment were confirmed by ¹H,¹H-COSY, HSQC, and HMBC experiments (*Fig. 1*). The C=C bond was located between C(20) and C(21), which was confirmed by HMBCs from C(20) to H_a -C(22) at δ (H) 1.98–2.00, Me(29) and Me(30), from C(21) to Me(30) and H_a -C(22), and from H–C(21) to C(17), C(19), and C(30). The CO group was located at C(3) due to HMBCs from the CO group at δ (C) 218.5 to CH₂(1), CH₂(2), H–C(5), Me(23), and Me(24). Two AcO groups were assigned to the C(11) and C(28) positions, respectively, based on the HMBCs from H–C(11) at δ (H) 4.92–4.94 to an AcO group with C(1[#]) at δ (C) 171.0 and from the CH₂O group with H_a–C(28) at δ (H) 3.77 and H_b–C(28) at δ (H) 4.27 to C(16), C(17), C(19) and the other AcO group with C(1^{*}) at δ (C) 171.2. The arabinopyranosyloxy residue was located at the C(6) position, based on the



Fig. 1. Key correlations from HMBC (H \rightarrow C), ¹H,¹H-COSY (---), and NOESY (H \rightarrow H) experiments of 1

HMBCs observed between H–C(6) at δ (H) 3.72 and the anomeric C-atom C(1') at δ (C) 101.0.

The relative configuration of **1** was elucidated by coupling constants, NOE, and NOESY experiments. The orientation of the (arabinopyranosyl) oxy residue at C(6)was assigned to be α due to the coupling constants of H–C(6) (δ (H) 3.72, br. dd, J = 10.5, 3.0) in the ¹H-NMR. This coupling pattern was similar to that found in other 6a-OH triterpenes such as missourin, which shows a *doublet* of *triplet* (J = 11.4, 7.2)pattern [21], in contrast to 6β -OH triterpenes (m, $W_{1/2} = 6$ Hz or br. s) [16][22]. In NOE experiments, irradiation of Me(25) lead to the enhancement of Me(24) (+10.38%), Me(26) (+10.48%), H_a-C(28) at $\delta(H)$ 3.77 (+7.09%) and H-C(11) (+5.90%) and irradiation of H-C(18) lead to the enhancement of Me(27) (+8.78%)and Me(29) (+6.68%). A NOESY experiment showed key correlations as shown (Fig. 1), especially significant correlations from Me(23) to $H_a - C(1)$ at 1.93–1.95 and H_a -C(2) at δ (H) 2.76, correlations from Me(24) to Me(25), H-C(6), and H_{β} -C(12) at $\delta(H)$ 1.74–1.82, correlations from Me(25) to Me(26), H_{β}-C(7) at $\delta(H)$ 1.99–2.01, and H-C(11), correlations from Me(26) to Me(24), H_{β} -C(1) at δ (H) 2.22-2.24, H-C(11), and H-C(19) at $\delta(H)$ 1.66–1.68, correlations from Me(27) to H-C(9), $H_a - C(12)$ at $\delta(H)$ 1.26–1.35, $H_a - C(16)$ at $\delta(H)$ 1.55–1.57, and H - C(18), correlations from Me(29) to Me(27), H_a -C(12) at $\delta(H)$ 1.26–1.35, and H-C(18), correlation between $H_a - C(16)$ at $\delta(H) 1.55 - 1.57$ and $H_a - C(22)$ at $\delta(H) 1.98 - 2.00$. If Me(25) was β -oriented as in all natural taraxastane-type triterpenoids, H-C(6), H-C(11), Me(24), Me(25), Me(26), and CH₂(28) should be β -oriented, but H-C(5), H-C(9), H-C(18), Me(23), Me(27), and Me(29) should be α -oriented. The signals of H_a - and H_β -C(15) were assigned by comparison with reported values [17][23].

Compound **2** was obtained as a crystalline powder. The HR-ESI-MS (positive ion mode) showed a *pseudo*-molecular-ion peak at m/z 748.4671 ($[M + NH_4]^+$, $C_{41}H_{66}NO_{11}^+$; calc. 748.4630), corresponding to the molecular formula $C_{41}H_{62}O_{11}$. The IR spectrum showed absorption bands at 3436 (OH), 1734 and 1711 (C=O) cm⁻¹. From ¹H- and ¹³C-NMR (*Table 1*), ¹H,¹H-COSY, HMBC (*Fig. 2*), and 1D-NOE data, the structure of **2** was elucidated as (6α ,11 α)-6-[(2-*O*-acetyl- β -D-xylopyranosyl)oxy]-3-oxotaraxast-20-ene-11,28-diyl diacetate.

The ¹H- and ¹³C-NMR spectral data of **2** were similar to those of **1** except for the signals of the sugar residue (*Table 1*), indicating that both compounds possess the same aglycone. The ¹H- and ¹³C-NMR spectra showed the sugar residue signals at δ (H) 4.61 (d, J = 6.3), 4.78 (t, J = 7.2), 3.54 (t, J = 8.2), 3.81–3.83 (m), 3.31 (dd, J = 11.7, 2.1), and 4.03 (dd, J = 11.4, 4.5); and at δ (C) 101.7, 73.7, 75.5, 70.3, and 64.3 (*Table 1*), which closely resembled those of β -xylopyranose (Xyl) [24]. The NOE experiment confirmed the sugar residue to be β -Xyl. Irradiation of H–C(1') led to the enhancement of H–C(3') (+4.57%) and irradiation of H–C(3') led to the enhancement of CH₂(5') at δ (H) 3.31 (+2.18%). The D-configuration of Xyl was assumed from biogenetic considerations. HMBCs, from H–C(2') to C(1'), C(3'), and C(1'') (δ (C) 171.4), revealed the substitution of HO–C(2') by an AcO group as in compound **1**.

Compound **3** was obtained as a colorless oil. The HR-ESI-MS (positive ion mode) showed a *pseudo*-molecular-ion peak at m/z 420.2379 ($[M + NH_4]^+$, $C_{23}H_{34}NO_6^+$; calc. 420.2381), corresponding to the molecular formula $C_{23}H_{30}O_6$. The IR spectrum showed absorption bands at 3514 (OH), 2931 (MeO), 1710 (conjugated C=C bond with



Fig. 2. Key correlations from HMBC ($H \rightarrow C$) and ¹H, ¹H-COSY (-) experiments of 2

MeO), 1601 and 1514 (aromatic moiety) cm⁻¹. From ¹H- and ¹³C-NMR (*Table 2*), ¹H,¹H-COSY, HMBC, and 1D-NOE data, the structure of **3** was elucidated as (1R,2R,4E)-1,5-bis(3,4-dimethoxyphenyl)-2-(methoxymethyl)pent-4-en-1-ol.

The ¹³C-NMR spectrum of **3** showed 23 signals (*Table 2*), assigned to twelve Catoms of two Ph groups, five MeO groups, an CH₂O group, an O–CH group, two olefinic and two non-oxygenated aliphatic C-atoms. The ¹H-NMR spectrum of **3** displayed the presence of two 1,3,4-trisubstituted Ph groups (δ (H) 6.92 (s), 6.82 (d, J = 7.0), 6.86 (d, J = 7.1), 6.86 (s), 6.80 (d, J = 7.2), and 6.85 (d, J = 7.2)), a (E)–C=C bond (δ (H) 6.28 (d, J = 15.6), 5.95 (dt, J = 15.6, 7.5)), one CH₂O (δ (H) 3.45 (dd, J = 9.3, 6.3), 3.59 (dd, J = 9.3, 3.6)), one CH₂ group at δ (H) 2.15 (ddd, J = 12.6, 7.5, 6.0) and 2.24 (ddd, J = 12.9, 7.5, 6.6), one O–CH group at δ (H) 3.36, 3.86, 3.87, 3.89, and 3.89. The ¹H- and ¹³C-NMR spectra of **3** were similar to those of morinol G (**7**) with the exception of the signals of an additional MeO group appearing at δ (H) 3.36 and δ (C) 59.1 [12].

The HMBC experiment showed correlations within two C_6-C_3 units. The first unit showed correlations from C(7)¹) at δ (C) 77.2 to H–C(6), H–C(8), H_a–C(9) and H_b–C(9); from C(8) at δ (C) 44.5 to H_a–C(9), H_b–C(9), H_a–C(9'), and H_b–C(9'); from CH₂(9) at δ (H) 3.45 and 3.59 to C(7), C(8), and a MeO C-atom at δ (C) 59.1, which indicated the presence of a MeO at C(9). The second unit showed correlations from C(7') at δ (C) 131.4 to H–C(6'), H–C(8'), H_a–C(9'), and H_b–C(9'); from C(8') at δ (C) 126.0 to H–C(7'), H_a–C(9'), and H_b–C(9'); from CH₂(9') at δ (H) 2.15 and 2.24 to C(7'), C(8'), C(8), and C(9), respectively. The two C₆–C₃ units are connected between C(8) and C(9') by a C–C bond. This was confirmed by ¹H,¹H-COSY correlations, between H–C(7) and H–C(8), from H–C(8) to CH₂(9) and CH₂(9'), and from H–C(8') to H–C(7') and CH₂(9').

The relative configuration of **3** was deduced from the coupling constant and an NOE experiment. The coupling constant between H-C(8) and H-C(7) was 6.6 Hz,

$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		$\delta(\mathrm{H})$		$\delta(C)$	
$C(1)$ 136.0 (s)135. $H-C(2)$ $6.92 (s)$ $6.83-6.85 (m)$ $109.4 (d)$ 110.0 $C(3)$ $148.2 (s)$ 148.2 $C(4)$ $148.2 (s)$ 148.2 $C(4)$ $148.9 (s)$ 148.2 $H-C(5)$ $6.82 (d, J=7.0)$ $6.82 (d, J=7.1)$ $111.1 (d)$ $H-C(6)$ $6.86 (d, J=7.1)$ $6.85-6.87 (m)$ $118.8 (d)$ $H-C(7)$ $4.69 (d, J=6.3)$ $4.18 (d, J=6.6)$ $77.2 (d)$ $H-C(8)$ $1.98-2.04 (m)$ $1.96-2.04 (m)$ $45.5 (d)$ $H-C(8)$ $1.98-2.04 (m)$ $1.96-2.04 (m)$ $45.5 (d)$ $CH_2(9)$ $3.45 (dd, J=9.3, 6.3, H_a)$, $3.25-3.40 (m, H_a)$, $74.2 (t)$ $3.59 (dd, J=9.3, 3.6, H_b)$ $3.55 (dd, J=9.6, 4.8, H_b)$ $C(1')$ $148.3 (s)$ $H-C(2')$ $6.86 (s)$ $6.83-6.85 (m)$ $108.5 (d)$ $C(3')$ $148.3 (s)$ $H-C(5')$ $6.80 (d, J=7.2)$ $6.80 (d, J=7.2)$ $110.7 (d)$ $H-C(6')$ $6.85 (d, J=7.2)$ $6.80 (d, J=7.2)$ $110.7 (d)$ $H-C(6')$ $6.85 (d, J=15.6)$ $6.21 (d, J=15.9)$ $131.4 (d)$ $H-C(8')$ $5.95 (dt, J=15.6, 7.5)$ $5.92 (dt, J=15.3, 7.2)$ $126.0 (d)$ $L^2(2')$ $2.15 (ddd, J=12.9, 7.5, 6.6, H_b)$ $2.17-2.21 (m, H_a)$, $32.2 (t)$ $MeO-C(3)$ $3.86 (s)$ $3.89 (s)$ $55.8 (q)$ $55.7 (q)$ $MeO-C(4')$ $3.89 (s)$ $3.89 (s)$ $55.8 (q)$ 5		3	4	3	4
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C(1)	_	_	136.0 (s)	135.6 (s)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(2)	6.92 (s)	6.83 - 6.85(m)	109.4(d)	110.0(d)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C(3)	_	_	148.2(s)	148.2(s)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C(4)	_	_	148.9(s)	148.9 (s)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(5)	6.82 (d, J = 7.0)	6.82 (d, J = 7.1)	111.1(d)	111.0(d)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(6)	6.86 (d, J = 7.1)	6.85 - 6.87 (m)	118.8(d)	120.1(d)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(7)	4.69(d, J = 6.3)	4.18(d, J = 6.6)	77.2(d)	83.1 (d)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(8)	1.98 - 2.04 (m)	1.96 - 2.04 (m)	45.5(d)	45.5(d)
$3.59 (dd, J = 9.3, 3.6, H_b)$ $3.55 (dd, J = 9.6, 4.8, H_b)$ $C(1')$ $130.5 (s)$ $130.$ $H-C(2')$ $6.86 (s)$ $6.83 - 6.85 (m)$ $108.5 (d)$ $108.$ $C(3')$ $148.3 (s)$ $148.$ $C(4')$ $148.9 (s)$ $148.$ $C(4')$ $148.9 (s)$ $148.$ $H-C(5')$ $6.80 (d, J = 7.2)$ $6.80 (d, J = 7.2)$ $110.7 (d)$ $110.$ $H-C(6')$ $6.85 (d, J = 7.2)$ $6.83 - 6.85 (m)$ $118.7 (d)$ $118.$ $H-C(7')$ $6.28 (d, J = 15.6)$ $6.21 (d, J = 15.9)$ $131.4 (d)$ 131.4 $H-C(8')$ $5.95 (dt, J = 15.6, 7.5)$ $5.92 (dt, J = 15.3, 7.2)$ $126.0 (d)$ $126.$ $CH_2(9')$ $2.15 (ddd, J = 12.6, 7.5, 6.0, H_a),$ $2.04 - 2.15 (m, H_a),$ $32.2 (t)$ $30.$ $2.24 (ddd, J = 12.9, 7.5, 6.6, H_b)$ $2.17 - 2.21 (m, H_b)$ $3.86 (s)$ $55.7 (q)$ $55.$ $MeO-C(3)$ $3.86 (s)$ $3.89 (s)$ $55.8 (q)$ $55.$ $MeO-C(4')$ $3.89 (s)$ $3.89 (s)$ $55.8 (q)$ $55.$ $MeO-C(4')$ $3.89 (s)$ $3.89 (s)$ $55.8 (q)$ $55.$ $MeO-C(7')$ $ 3.21 (s)$ $ 50.1 (s)$	CH ₂ (9)	$3.45 (dd, J = 9.3, 6.3, H_a),$	$3.25 - 3.40 (m, H_a),$	74.2(t)	71.5(t)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$,	$3.59 (dd, J = 9.3, 3.6, H_{\rm b})$	$3.55 (dd, J = 9.6, 4.8, H_{\rm b})$		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C(1')	_	_	130.5(s)	130.8(s)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(2')	6.86(s)	6.83 - 6.85(m)	108.5(d)	108.4(d)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	C(3')	_	_	148.3(s)	148.3 (s)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	C(4')	_	_	148.9(s)	148.9 (s)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(5')	6.80 (d, J = 7.2)	6.80 (d, J = 7.2)	110.7(d)	110.5(d)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(6')	6.85(d, J = 7.2)	6.83 - 6.85(m)	118.7(d)	118.8(d)
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	H-C(7')	6.28 (d, J = 15.6)	6.21 (d, J = 15.9)	131.4(d)	131.0(d)
$\begin{array}{c} \mathrm{CH}_2(9') & 2.15 \ (ddd, J = 12.6, 7.5, 6.0, \mathrm{H}_a), & 2.04 - 2.15 \ (m, \mathrm{H}_a), & 32.2 \ (t) & 30. \\ 2.24 \ (ddd, J = 12.9, 7.5, 6.6, \mathrm{H}_b) & 2.17 - 2.21 \ (m, \mathrm{H}_b) & \\ \mathrm{MeO} - \mathrm{C}(3) & 3.86 \ (s) & 3.86 \ (s) & 55.7 \ (q) & 55. \\ \mathrm{MeO} - \mathrm{C}(4) & 3.89 \ (s) & 3.87 \ (s) & 55.8 \ (q) & 55. \\ \mathrm{MeO} - \mathrm{C}(4') & 3.89 \ (s) & 3.89 \ (s) & 55.8 \ (q) & 55. \\ \mathrm{MeO} - \mathrm{C}(4') & 3.89 \ (s) & 3.89 \ (s) & 55.8 \ (q) & 55. \\ \mathrm{MeO} - \mathrm{C}(4') & 3.89 \ (s) & 3.89 \ (s) & 55.8 \ (q) & 55. \\ \mathrm{MeO} - \mathrm{C}(4') & 3.89 \ (s) & 3.89 \ (s) & 55.8 \ (q) & 55. \\ \mathrm{MeO} - \mathrm{C}(7) & - & 3.21 \ (s) & - & 56. \\ \mathrm{MeO} - \mathrm{C}(9) & 2.26 \ (s) & - & 501 \ (s) & - & 501 \ (s) \\ \end{array}$	H-C(8')	5.95 (dt, J = 15.6, 7.5)	5.92 (dt, J = 15.3, 7.2)	126.0(d)	126.8(d)
$2.24 (ddd, J = 12.9, 7.5, 6.6, H_b)$ $2.17 - 2.21 (m, H_b)$ MeO - C(3) $3.86 (s)$ $3.86 (s)$ MeO - C(4) $3.89 (s)$ $55.7 (q)$ MeO - C(3') $3.87 (s)$ $55.7 (q)$ MeO - C(4') $3.89 (s)$ $55.7 (q)$ MeO - C(4') $3.89 (s)$ $55.7 (q)$ MeO - C(4') $3.89 (s)$ $55.8 (q)$ MeO - C(4') $3.89 (s)$ $55.8 (q)$ MeO - C(7) - $3.21 (s)$ MeO - C(0) $2.26 (s)$ $52.1 (s)$	CH ₂ (9')	$2.15 (ddd, J = 12.6, 7.5, 6.0, H_a),$	$2.04 - 2.15 (m, H_a),$	32.2(t)	30.5(t)
MeO-C(3) $3.86(s)$ $3.86(s)$ $55.7(q)$ $55.$ $MeO-C(4)$ $3.89(s)$ $3.89(s)$ $55.8(q)$ $55.$ $MeO-C(3')$ $3.87(s)$ $3.87(s)$ $55.7(q)$ $55.$ $MeO-C(4')$ $3.89(s)$ $3.89(s)$ $55.8(q)$ $55.$ $MeO-C(7)$ $ 3.21(s)$ $ 56.$ $MeO-C(0)$ $2.25(s)$ $ 50.1(s)$		$2.24 (ddd, J = 12.9, 7.5, 6.6, H_{\rm b})$	$2.17 - 2.21 (m, H_{\rm b})$		
MeO-C(4) $3.89(s)$ $3.89(s)$ $55.8(q)$ $55.$ $MeO-C(3')$ $3.87(s)$ $3.87(s)$ $55.7(q)$ $55.$ $MeO-C(4')$ $3.89(s)$ $3.89(s)$ $55.8(q)$ $55.$ $MeO-C(7)$ $ 3.21(s)$ $ 56.$ $MeO-C(0)$ $2.25(s)$ $59.1(s)$ $ 50.1(s)$	MeO-C(3)	3.86 (s)	3.86 (s)	55.7(q)	55.7(q)
MeO-C(3') $3.87(s)$ $5.7(q)$ $55.$ $MeO-C(4')$ $3.89(s)$ $3.89(s)$ $55.8(q)$ $55.$ $MeO-C(7)$ $ 3.21(s)$ $ 56.$ $MeO-C(0)$ $2.25(s)$ $ 50.1(s)$ $-$	MeO-C(4)	3.89 (s)	3.89(s)	55.8(q)	55.8(q)
MeO-C(4') $3.89(s)$ $55.8(q)$ $55.$ $MeO-C(7)$ - $3.21(s)$ - $56.$ $MeO-C(7)$ - $52.5(s)$ - $56.$	MeO-C(3')	3.87 (s)	3.87(s)	55.7(q)	55.7(q)
MeO - C(7) - 3.21 (s) - 56.	MeO-C(4')	3.89(s)	3.89(s)	55.8(q)	55.9(q)
$M_{-0} = 0$	MeO-C(7)	_	3.21(s)	-	56.8 (q)
MeO-C(9) = 5.56(8) = 5.55(8) = 59.1(q) = 58.1(q) = 58.	MeO-C(9)	3.36 (s)	3.35 (s)	59.1 (q)	58.9 (q)

Table 2. ¹*H*- and ¹³*C*-*NMR* (DEPT) *Data*^a) of **3** and **4** (CDCl₃, δ in ppm, *J* in Hz)¹)

which suggested that the configuration is *threo* in comparison with the values reported for *threo* (J = 6.9) and *erythro* (J = 4.1) configurations [25]. In the NOE experiment, irradiation of H–C(8) lead to the enhancement of H–C(7) (+3.15%). The CD spectrum of **3** showed a negative excition chirality by negative excition coupling (λ ([θ]) = 256.8 (+125.7), 265.5 (-3660)).

Compound **4** was obtained as a colourless oil. The HR-ESI-MS (positive ion mode) showed a *pseudo*-molecular-ion peak at m/z 434.2532 ($[M + NH_4]^+$, $C_{24}H_{36}NO_6^+$; calc. 434.2537), corresponding to the molecular formula $C_{24}H_{32}O_6$. Its IR spectrum showed absorption bands at 2927 (MeO), 1713 (conjugated C=C bond with MeO), 1652, 1597, and 1513 (aromatic moiety) cm⁻¹. From ¹H- and ¹³C-NMR (*Table 1*), ¹H,¹H-COSY, HMBC (*Fig. 3*), and 1D-NOE data, compound **4** was elucidated as 1,1'-[(1E,4R,5R)-5-methoxy-4-(methoxymethyl)pent-1-ene-1,5-diyl]bis(3,4-dimethoxybenzene).

^a) The assignments of the signals of **3** and **4** were performed by ¹H,¹H-COSY, HMBC, and NOE experiments.

¹⁾ Arbitrary atom numbering. For systematic names, see Exper. Part.



Fig. 3. Key correlations from HMBC $(H \rightarrow C)$ experiments of 4^{1})

The ¹H- and ¹³C-NMR spectral data of **4** were very similar to those of **3** with the exception that signals of an additional MeO group appeared at $\delta(H)$ 3.21 and $\delta(C)$ 56.8. The only other difference in the ¹³C-NMR spectral data between **3** and **4** occured for the C-atoms C(7) and C(9)¹). In the ¹³C-NMR spectrum, the signal of C(7) was shifted downfield ($\delta(C)$ 83.1) and that of C(9) upfield because of the additional MeO group located at C(7) in **4** (*Table 2*). In the ¹H-NMR spectrum, the signals of H–C(7) ($\delta(H)$ 4.18 (d, J = 6.3)), H–C(9), and H–C(9') were all shifted upfield in comparison to the corresponding signals in **3**. The HMBC between the MeO group at $\delta(C)$ 56.8 and H–C(7) also confirmed the position of the MeO substituent. In the NOE experiment, irradiation of H–C(8) lead to the enhancement of H–C(7) (+2.98%), which indicated that the configuration of C(7) and C(8) was *threo*. The CD spectrum of **4** showed a negative excition chirality by negative excition coupling ($\lambda([\theta]) = 266.5$ (-19.6), 269.5 (-1700)). The CD spectra of **3** and **4** revealed that both compounds possess the same absolute configurations at C(7) and C(8), respectively [25].

Phytochemical studies on the genus *Morina* mainly focused on two species, *M. chinensis* and *M. nepalensis* var. *alba*. Flavonoid glycosides, neolignans, sesquineolignans, phenylpropanoid derivatives, and ursane-type saponins were isolated from *M. chinensis*. Acylated flavonoid glycosides, caffeoylquinic acids, oleanane- and ursane-type saponins were isolated from *M. nepalensis* var. *alba*. In this work, we isolated lignans, neolignans, taraxastane- and ursane-type triterpenes from *M. kokonorica*. Except for **7**, compounds were all isolated from the genus *Morina* for the first time. To our knowledge, compounds **1** and **2** are the first taraxastane-type triterpenes glycosylated at C(6).

Experimental Part

General. Column chromatography (CC): silica gel (SiO₂; 200–300 mesh, Qingdao Marine Chemical Factory, P. R. China); SiO₂ GF₂₅₄ (10–40 µm, Qingdao Marine Chemical Factory), detection at 254 nm UV light or by heating after spraying with 5% H₂SO₄ in EtOH (ν/ν). M.p.: Kofler melting point apparatus; uncorrected. Optical rotations: Perkin-Elmer-341 polarimeter. CD Spectra: Jasco-J-810 circular dichroism spectropolarimeter, λ ([θ]) in nm. UV Spectra: Shimadzu spectrometer UV-260, λ_{max} (log ε) in nm. IR Spectra: Nicolet NEXUS-670 FT-IR spectrometer, in cm⁻¹. NMR Spectra: Varian Mercury plus-300 spectrometer at 300 (¹H-NMR) and 75 MHz (¹³C-NMR), δ in ppm, J in Hz, Me₄Si (TMS) as internal standard. HR-ESI-MS: Bruker Daltonics APEX-II mass spectrometer.

Plant Material. The whole plant of *M. kokonorica* was collected in the south of Gansu Province, P. R. China, in September 2005. It was identified by Prof. *Guo-Liang Zhang*, Department of Life Science, Lanzhou University. A voucher specimen (No. Mk20050906) has been deposited with the State Key Laboratory of Applied Organic Chemistry, Lanzhou University.

Extraction and Isolation. The powder of the air-dried whole plant of *M. kokonorica* (4 kg) was extracted with MeOH (251 each time) at *ca.* 5° for four times, 7 d each. After concentration of the extracts under reduced pressure, the residue (643 g) was suspended in H₂O (5.51) and then extracted with CHCl₃ (4 × 3.51). The CHCl₃ extract (213.5 g) was subjected to CC (SiO₂, 1900 g) by gradient elution with petroleum ether (PE; $60-90^{\circ}$)/acetone (100:1 to 1:1) and finally washed with MeOH, to give 20 fractions (*Fr.* 1–*Fr.* 20). *Fr.* 3 (16 g; PE/acetone 20:1) was resubjected to CC (SiO₂; PE/acetone 50:1 to 10:1) and further separated by CC (PE/ACOEt 10:1), to obtain **5** (7 mg). *Fr.* 6 (4.0 g, PE/acetone 5:1) was subjected to CC (SiO₂; PE/acetone 8:1 to 5:1) and further separated by CC (PE/AcOEt 3:1) to give **6** (8 mg) and **4** (7 mg). *Fr.* 12 (5.1 g; PE/acetone 3:1) was subjected to CC (SiO₂; PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/AcOEt 5:1 to 1:1) and further separated by CC (PE/acetone 2:1) was subjected to CC (SiO₂; PE/AcOEt 3:1 to 1:1) and further separated by CC (PE/acetone 2:1) was subjected to CC (PE/acetone 2:1) to obtain **10** (6 mg), **7** (10 mg), and **1** (22 mg). *Fr.* 16 (3.1 g; PE/acetone 1:1) was subjected to CC (SiO₂; PE/AcOEt 1:1) and further separated by CC (PE/acetone 2:1) to obtain **10** (6 mg).

(6a,11a)-6-[(2-O-Acetyl-a-L-arabinopyranosyl)oxy]-3-oxotaraxast-20-ene-11,28-diyl Diacetate (1). White crystalline powder (acetone). M.p. 235–236°. $[a]_D^{20} = +15$ (c = 1.0, acetone). IR (KBr): 3435, 2962, 2929, 1735, 1639, 1564, 1456, 1369, 1239, 1089, 1053, 1029, 967, 773. ¹H- and ¹³C-NMR (DEPT): see Table 1. HR-ESI-MS: 748.4633 ($[M + NH_4]^+$, $C_{41}H_{66}NO_{11}^+$; calc. 748.4630).

 $(6\alpha,11\alpha)$ -6-[(2-O-Acetyl- β -D-xylopyranosyl)oxy]-3-oxotaraxast-20-ene-11,28-diyl Diacetate (2). White crystalline powder (acetone). M.p. $233-234^{\circ}$. $[\alpha]_D^{20} = +12$ (c = 1.1, acetone). IR (KBr): 3436, 2960, 2930, 2876, 1734, 1711, 1514, 1458, 1368, 1239, 1080, 1033, 984. ¹H- and ¹³C-NMR (DEPT): see Table 1. HR-ESI-MS: 748.4671 ($[M + NH_4]^+$, $C_{41}H_{66}NO_{11}^+$; calc. 748.4630).

(IR,2R,4E)-1,5-Bis(3,4-dimethoxyphenyl)-2-(methoxymethyl)pent-4-en-1-ol (3). Colorless oil. $[\alpha]_D^{20} = +1$ (c = 0.30, MeOH). CD (MeOH): 256.8 (+125.7), 265.5 (-3660), 274.8 (-1681). UV (MeOH): 262 (2.72), 225 (2.98). IR (film): 3514, 2997, 2931, 2835, 1710, 1601, 1514, 1462, 1417, 1358, 1263, 1235, 1139, 1027, 967, 859, 812, 765. ¹H- and ¹³C-NMR (DEPT): see *Table 2*. HR-ESI-MS: 420.2379 ($[M + NH_4]^+$, C₂₃H₃₄NO₆⁺; calc. 420.2381).

1,1'-[(1E,4R,5R)-5-Methoxy-4-(methoxymethyl)pent-1-ene-1,5-diyl]bis(3,4-dimethoxybenzene) (4). Colorless oil. $[a]_D^{2D} = +2$ (c = 1.00, MeOH). CD (MeOH): 266.5 (-19.6), 269.5 (-1700). UV (MeOH): 267 (3.36), 231 (3.38). IR (film): 2927, 2835, 1713, 1652, 1597, 1513, 1461, 1417, 1357, 1263, 1235, 1138, 1099, 1028, 966, 859, 809, 764. ¹H- and ¹³C-NMR (DEPT): see *Table 2*. HR-ESI-MS: 434.2532 ($[M + NH_4]^+$, C₂₄H₃₆NO₆⁺; calc. 434.2537).

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REFERENCES

- 'Delectis Florea Reipublicea Popularis Sinicae Agendae Academiae Sinicea Edita', in 'Flora Reipublicae Popularis Sinicea', Tomus, Science Press, Beijing, China, 1986, Vol. 73(1), pp. 44–51.
- [2] Academia Sinica, 'North-Western Plateau Institute of Biology', in 'The Medico-Flora of Tibetan', Ed. Y. C. Yang, Qinghai People's Press, Xining, 1991, pp. 200–201.
- [3] B.-N. Su, Y. Takaishi, H.-Q. Duan, B. Chen, J. Nat. Prod. 1999, 62, 1363.
- [4] R. W. Teng, H. Y. Xie, D. Z. Wang, C. R. Yang, Magn. Reson. Chem. 2002, 40, 603.
- [5] K. I. Alimov, K. K. Khalmatov, I. A. Kharlamov, M. T. Ikramov, Khim. Prir. Soedin. 1981, 2, 248.
- [6] M. Ali, K. K. Bhutani, J. Gupta, Pharmakeutike 1995, 8, 114.
- [7] Y.-F. Wang, Zh.-H. Wu, X.-Y. Zhou, J.-J. Ye, Q.-F. Qian, Acta Bot. Sin. 1986, 28, 566.
- [8] K. H. C. Baser, M. Kurkcuoglu, J. Essent. Oil Res. 1998, 10, 117.
- [9] B. D'Abrosca, A. Fiorentino, P. Monaco, P. Oriano, S. Pacifico, Food Chem. 2006, 98, 285.
- [10] R. K. Hota, M. Bapuji, *Phytochemistry* **1993**, *32*, 466.
- [11] S. Begum, Farhat, I. Sultana, B. S. Siddiqui, F. Shaheen, A. H. Gilani, J. Nat. Prod. 2000, 63, 1265.

- [12] B.-N. Su, Y. Takaishi, T. Kusumi, Tetrahedron 1999, 55, 14571.
- [13] B. Vermes, O. Seligmann, H. Wagner, Phytochemistry 1991, 30, 3087.
- [14] S. Ferreira Fonseca, J. de Paiva Campello, L. E. S. Barata, E. A. Rúveda, *Phytochemistry* 1978, 17, 499.
- [15] L.-K. Sy, G. D. Brown, Phytochemistry 1999, 50, 781.
- [16] M. L. Flagg, S. Valcic, G. Montenegro, M. Gomez, B. N. Timmermann, *Phytochemistry* 1999, 52, 1345.
- [17] W. F. Reynolds, S. McLean, J. Poplawski, R. G. Enriquez, L. I. Escobar, I. Leon, *Tetrahedron* 1986, 42, 3419.
- [18] X.-F. He, X.-N. Wang, C.-Q. Fan, L.-S. Gan, S. Yin, J.-M. Yue, Helv. Chim. Acta 2007, 90, 783.
- [19] J. Youkwan, P. Srisomphot, S. Sutthivaiyakit, J. Nat. Prod. 2005, 68, 1006.
- [20] I. Kitagawa, K. S. Im, Y. Fujimoto, Chem. Pharm. Bull. 1977, 25, 800.
- [21] S. M. Wong, Y. Oshima, J. M. Pezzuto, H. H. S. Fong, N. R. Farnsworth, J. Pharm. Sci. 1986, 75, 317.
- [22] R. Aquino, F. De Simone, F. F. Vincieri, C. Pizza, E. Gaćs-Baitz, J. Nat. Prod. 1990, 53, 559.
- [23] H.-D. Chen, S.-P. Yang, S.-G. Liao, C.-R. Zhang, J.-M. Yue, Helv. Chim. Acta 2006, 89, 1971.
- [24] X.-J. Gu, Y.-B. Li, P. Li, S.-H. Qian, J.-F. Zhang, Helv. Chim. Acta 2007, 90, 72.
- [25] T. Miyase, A. Ueno, N. Takizawa, H. Kobayashi, H. Oguchi, Chem. Pharm. Bull. 1987, 35, 3713.

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